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COMPLETE SPECIFICATION

Detergent Mixtures comprising Anionic Detergents and Cation-Active Compounds

We, REWO CHEMISCHE FABRIK G.M.B.H., a German Body Corporate, of Steinau, Kreis Schlüchtern, Germany, do hereby declare the invention, for which we pray that a patent may be granted to us, and the method by which it is to be performed, to be particularly described in and by the following statement:—

The present invention is concerned with detergent mixtures comprising anionic deter-

gents and cation-active compounds.

Anionic compositions are readily available and used for a whole host of pharmaceutical, cosmetic, hygienic, detergent or cleansing purposes. They may, for example, take the form of washing and cleansing compositions for textiles, e.g. blankets and carpets; solutions for cleansing medical equipment, food containers and cutlery; hair preparations and shampoos; disinfectants; and deodorants. The anionic compositions are prepared from anionic mixtures which are diluted with water to provide aqueous solutions to which additives, e.g. germicides, emulsifiers, sequestrants, colourants and propellants may be added. One form of mixture comprises anionic detergents and cation-active compounds but such mixtures have the drawbacks that clear aqueous solutions cannot be obtained, the solutions react with turbid amorphous or crystalline masses, e.g. soaps, and above all that unless the very minimum of cation-active compound is used, there is a tendency for the detergent to be precipitated. Because of this. such mixtures are usually used to prepare pastes, soaps, syndets or powders. It is possible to coat the cation-active compounds with inorganic electrically neutral salts in a preliminary stage but the resultant mixture only provides clear aqueous solutions when the concentration of anionic detergent is kept below 0.1% by weight. It is also possible to

secure clear aqueous solutions of higher detergent concentration where the detergent is a sulphated or sulphonated compound, but only if very high concentrations of non-ionic substances are included. All in all, therefore, these mixtures suffer from serious disadvantages.

It has now been discovered that if only certain anionic detergents and certain cationic compounds are employed in a certain ratio range, the disadvantages mentioned are minimised, if not avoided completely. This invention therefore makes it possible to prepare clear aqueous solutions containing practically any concentration of anionic detergent which solutions, moreover, exhibit a high degree of viscosity even at low levels of dilution.

According to this invention, there is therefore provided a mixture for use as or in the preparation of solutions for any of the purposes hereinbefore specified, comprising a multi-functional anionic detergent and a cation-active compound, the proportion of the latter to the former not exceeding the molar

ratio of

n - :

where n and m represent the number of functions of the anionic detergent and cationactive compound, respectively.

The term "functions" in this connection means the number of reactive groups present in the compound. For example maleic acid having two reactive carboxylic groups is bifunctional and a sulphosuccinic acid having two reactive carboxylic acid groups and one reactive sulphate group in the molecule is tri-

functional.

In the case of a bifunctional anionic detergent, the proportion of cationic compound (for

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example, a quaternary ammonium compound), may be up to a molar ratio of 1:1 and using a tri-functional anionic detergent, the proportion may be in a molar ratio of up to 1:2.

Preferred bifunctional anionic detergents include the sulphosuccinic acid derivatives of the general formula

in which R' represents an alkyl residue having
at least 6 carbon atoms, Y represents a hydrocarbon chain optionally substituted with nitrogen or oxygen or both, either of X and Z
represents a hydrogen atom and the other is
the group —SO₂M, and M is a pharmateutically or cosmetically acceptable cation e.g.
a metal, amine, alkanolamine or ammonium.
Other anionic detergents include:—

sulpho-fatty acids, for example a-sulpho-fatty-acid;

compounds with two sulpho-groups; compounds with one sulpho-group and one carboxyl group;

compounds with two carboxyl groups in the molecule;

phosphonic acids with an alkyl residue and two acid functions;

and alkylphosphoric acid esters with two acid functions.

Appropriate cation-active compounds include the so-called "onium compounds"; for example, the ammonium, phosphonium, pyridinium and sulphonium compounds. Compounds preferred for commercial availability and manufacturing capabilities include the quaternary ammonium compounds of the formula

$$\begin{bmatrix} R & R & R^{(+)} \\ R & R & R^{(+)} \end{bmatrix} \chi^{(-)}$$

(II)

in which X is an anion and R represents an alkyl group. Preferred anions for substituent X include the halides, the saccharinate anions, the phthalimidate anions, $HSO_4(-)$ and the residue of an acid sulphimide. One of substituents R is desirably a long-chain alkyl residue having from 4 to 22 carbon atoms.

The mixtures may contain non-ionic com-

pounds and the sual washing agent additives for example phosphates, silicates and inorganic salts. As a rule they can be mixed with certain monofunctional anionic compounds without precipitation of insoluble components.

One advantage of the mixture of this invention is that a detergent compound usually considered to be insoluble can readily be formulated into an aqueous solution. If, for example, a sulphosuccinate which is soluble in water with difficulty in a 40% aqueous solution is mixed with an alkyldimethylbenzylammonium chloride in equimolar proportions, the result is a thick paste which requires dilution to provide a clear solution. (The sulphosuccinate alone would have yielded the same product). On the other hand, if one instead follows the procedures of this invention and mixes a water-soluble sulphossucinate with a waterinsoluble quaternary ammonium salt (for example, an alkyldimethylbenzylammonium saccharinate) in equimolar proportions, the result is a clear aqueous solution, even at a concentration of 40%.

A further advantage of the invention is that the solutions obtained are clear at high concentrations of detergent and another is that despite their anionic behaviour the solutions fully retain the properties of the cationic ingredient. Moreover, the use of "onium" compounds increases the viscosity considerably, so that, for example, even solutions with detergent contents of 10 to 5%, still show a high viscosity. Further, the wetting power and also the calcium soap dispersing capacity is increased to a considerable extent without reduction of the cleansing or washing power and of the lathering capacity.

Evaluations have also shown that materials cleaned by solutions of this invention are less susceptible to re-soiling due to electrostatic charge. Other evaluations have shown that the degreasing effect on skin and hair characteristic of the usual anion-active detergent compositions which contain cation-active material is diminished to a considerable extent.

Further advantages are to be seen in that the activity of a cationic compound in a mixture of this invention is uninhibited by the presence of the anionic ingredient. For example, if the cationic ingredient is a quaternary ammonium compound, one can still take advantage of its disinfectant and antimycotic properties. Moreover, solutions prepared from mixtures of this invention can be used for chemico-technical purposes, for example, lending water repellence to natural or synthetic fibres or other substances and rendering synthetic fibres and synthetic resins ("plastics") anti-static.

In order that this invention may be well understood, we now given some evaluations to show the properties of the mixtures of this invention and then formulations to prepare various types of solutions from them.

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EVALUATIONS.

I. Evaluation to show compatibility of cation compounds with anionic tensides. An aqueous washing solution containing

20% of anionic tenside was prepared and a quaternary compound added in a concentration of 20% of the detergent content, in accordance with the following Table.

	Appearance of solution when cation compound specified below added	
Anion-active tenside present in solution	Alkyldimethylbenzyl- ammonium chloride	Alkyldimethylbenzyl- ammonium saccharinate
Lauryl alcoholdiglycolether sulphate, Na salt	turbid	slightly turbid
Laurylalcohol sulphate, triethanol- amine salt	turbid	turbid
Laurylalcoholtriglycolether sulpho- succinic acid ester, Na salt	clear	clear
Laurylalcoholdiglycolether sulpho- succinic acid ester, Na monoethanol- amine salt	clear	clear

10 II. Evaluation to determine the bactericidal effect of a solution according to the invention.

1% of each quaternary product identified below was dissolved in an aqueous solution of 15 12% of the sodium salt of lauryltriglycolether sulphosuccinic acid ester. The content of

quaternary substance relative to the content of anionic detergent was about 9%, the total content of detergent being about 1%. Each solution was tested against Staphylococcus aureus according to the round filter method and the following results obtained

TABLE II

Active ingredient	Restriction of growth	
Trave Infrared	Zone 1.	Zone 2.
Cleansing agent alone	0	0
with 1% of alkyldimethyl benzylammonium saccharinate	7	7
with 1% of alkyltri methylammonium saccharinate	3	3
with 1% of alkyldimethyl benzylammonium phthalimidate	. 7	7

FORMULATIONS.

1. A Viscous liquid shampoo having anti-

dandruff and anti-static effect:

16 parts of lauryltriglycolether sulphosuccinic acid ester, Na salt.

1 part of alkyldimethylbenzylammonium saccharinate

4 parts of lauric acid diethanolamide

2 parts of common salt and

77 parts of water were admixed with gentle heating to provide a clear hair shampoo of agreeable viscosity.

2. A disinfectant:

16 parts of lauryltriglycolether sulphosuccinic acid ester, Na sait.

1 part of alkyldimethylbenzylalkonium chloride

4 parts of nonylphenolpolyglycolether

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were admixed with heating to provide a clear viscous liquid cleansing agent.

3. A disinfectant:

16 parts lauryldiglycolether sulphosuccinic acid ester, Na monoethanolamine salt

1 part alkyldimethylbenzylammonium phthalimidate

4 parts coconut fatty acid diethanolamide

5 parts common salt and

74 parts water

were admixed with heating and gave a clear viscous liquid rinsing agent exhibiting a dis-15 infecting action.

4. A liquid cleansing agent having disinfectant properties: -

29 parts of sulphosuccinic acid semi-ester of fatty alcohol polyglycolether, Na salt

20 1 part alkyldimethylbenzylammonium saccharinate

70 parts water

100 parts of a liquid cleansing agent containing a disinfectant

5. A liquid rinsing agent having disinfectant properties:-

20 parts of Na-alkylaryl sulphonate.

5 parts fatty acid diethanolamide

5 parts fatty alcohol diglycolether sulphate, Na salt

4.5 parts sulphosuccinic acid semi-ester of fatty alcohol polyglycolether, Na salt

0.5 parts alkyldimethylbenzylammonium chloride

35 Remainder water to 100

6. A liquid car-wash:

10 parts of sulphosuccinic acid semi-ester of fatty alcohol polyglycolether, Namonoethanolamine salt,

10 parts of alkyldimethylhydroxyethyl-40 ammonium chloride Remainder water to 100

7. A liquid water-repellent and anti-static

cleansing agent:

15 parts of sulphosuccinic acid semi-ester of fatty alcohol polyglycolether, Na-NH, salt

10 parts alkylhydroxyethylimidazolinium chloride

50 Remainder water to 100.

8. A disinfectant for personal use:

10 parts sulphosuccinic acid semi-ester of fatty alcohol polyglycolether, sodium salt

parts fatty acid diethanolamide

5 parts alkyldimethylbenzylammonium 55 phthalimidate

Remainder water to 100. 9. A liquid de-odourizing foam bath:

30 parts sulphosuccinic acid semi-ester of fatty acid alkylolamide, Na salt.

6 parts fatty acid alkylolamide

parts alkyldimethylbenzylammonium saccharinate

Remainder water to 100.

10. An anti-static wash paste: -

30 parts sulphosuccinic acid semi-ester of fatty acid alkylolamide, Na salt

parts fatty acid monoethanolamide

parts fatty alcohol sulphate

2 parts alkyltrimethylammonium sulphimidate

Remainder water to 100.

WHAT WE CLAIM IS:—

1. A mixture comprising a multi-functional anionic detergent and a cation-active compound, the proportion of the latter to the former not exceeding the molar ratio of

where n and m represent the number of functions of the anionic detergent and cationactive compound, respectively.

2. A mixture as claimed in claim 1, which the anionic detergent is bifunctional and the proportion of anionic detergent to cationic compound is up to the molar ratio of 1:1.

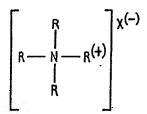
3. A mixture as claimed in claim 1, in which the anionic detergent is trifunctional and the proportion of anionic detergent to cationic compound is up to the molar ratio of 1:2.

4. A detergent mixture comprising a multifunctional anionic detergent and a cationactive compound selected from the class of socalled onium compounds, the proportion of the latter to the former in the composition not exceeding the molar ratio of

where n and m represent the number of functions of the anionic detergent and cationactive-compound respectively.

5. A mixture as claimed in claim 4 in which the onium compound is an ammonium, phosphonium, pyridinium or sulphonium compound.

6. A mixture as claimed in claim 5, in which the onium compound is a quaternary ammonium compound of the general formula



in which R represent an alkyl group and X is an anion.

7. A mixture as claimed in claim 6, in which X represents an halide, the phthalimi-

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date, saccharinate, HSO,— or acid sulphimide anion.

8. A mixture as claimed in claim 6, in which one of substituents R represents a long chain alkyl residue having from 4 to 22 carbon atoms.

9. A mixture comprising a bi-functional anionic detergent of the general formula

in which R' represents an alkyl residue having at least 6 carbon atoms, Y represents a hydrocarbon chain optionally substituted with oxygen or nitrogen or both, either of X and Z represents a hydrogen atom and the other is the group SO₃M, and M is a pharmaceutically or cosmetically acceptable cation; and a cationactive compound, the proportion of the latter to the former in the composition not exceeding the molar ratio of

$$\frac{n-1}{m}$$

where n and m represent the number of functions of the anionic detergent and cationactive compounds, respectively.

10. A mixture as claimed in claim 9, in which substituent M of the general formula represents a metal, amine, alkanolamine or ammonium.

11. A mixture as claimed in claim 9, in which the anionic detergent is a sulpho-fatty acid, a compound with two sulpho-groups or with one sulpho-group and a carboxyl-group or with two carboxyl groups in the molecule, a phosphonic acid with an alkyl residue and two acid functions or an alkylphosphoric acid with two acid functions.

12: A mixture comprising a multifunctional anionic sulphosuccinic detergent of the general formula

40 (in which R' represents an alkyl residue having at least 6 carbon atoms, Y represents a hydrocarbon chain optionally substituted with oxygen or nitrogen or both, either of X and Z represents a hydrogen atom and the other is the group —SO₂M, and M is a pharmaceutically or cosmetically acceptable cation), and a

cation quaternary ammonium compound of the general formula

$$\begin{bmatrix} R & R & R^{(+)} \\ R & R & R^{(+)} \end{bmatrix}^{\chi(-)}$$

(in which R is an alkyl group and X is an anion), the proportion of the latter to the former not exceeding the molar ratio of

$$\frac{n-1}{m}$$

where n and m represent the number of functions of the anionic sulphosuccinic detergent and cation quaternary ammonium compound respectively.

13. A mixture as claimed in claim 12, in which the anionic detergent is bifunctional and the proportion of anionic detergent to cationic compound is up to the molar ratio of

14. A mixture as claimed in claim 12, in which the anionic detergent is trifunctional and the proportion of anionic detergent to cationic compound is up to the molar ratio of

15. A mixture as claimed in any of the preceding claims, in which the anionic detergent is a lauryl alcohol derivative.

16. A mixture as claimed in any of the preceding claims, in which the anionic detergent is a lauryl alcohol derivative of a sulphosuccinic acid.

17. A mixture as claimed in any of the preceding claims, in which the cationic compound is an alkyldimethylbenzylammonium compound.

18. A mixture as claimed in any of the preceding claims, substantially as herein described.

An aqueous solution of a mixture as claimed in any of the preceding claims.
 A cleaning composition formed from a

mixture as claimed in any claims 1 to 18.

21. A cosmetic composition formed from a mixture as claimed in any of claims 1 to 18.

22. A composition formed from a mixture comprising a multifunctional anionic detergent and a cation compound substantially as herein described with reference to any of formulations 1 to 10.

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